

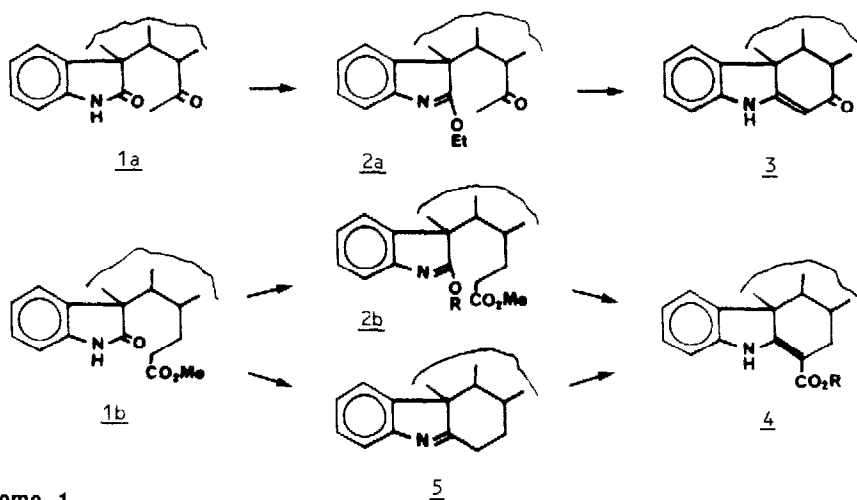
## CYCLIZATION OF OXINDOLIC METHYLKETONES WITH ACID : A RAPID SYNTHESIS OF ( $\pm$ )-ASPIDOFRACTININE

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**Summary :** The oxindolic methylketone 1 was cyclized in one step to the hexacyclic ketolactam 3 with acid. ( $\pm$ )-Aspidofractinine 23 was prepared by sequential reduction of 3. The model oxindolic methylketone 11, when reacted with acid, gave compounds 12 and 17a,b.

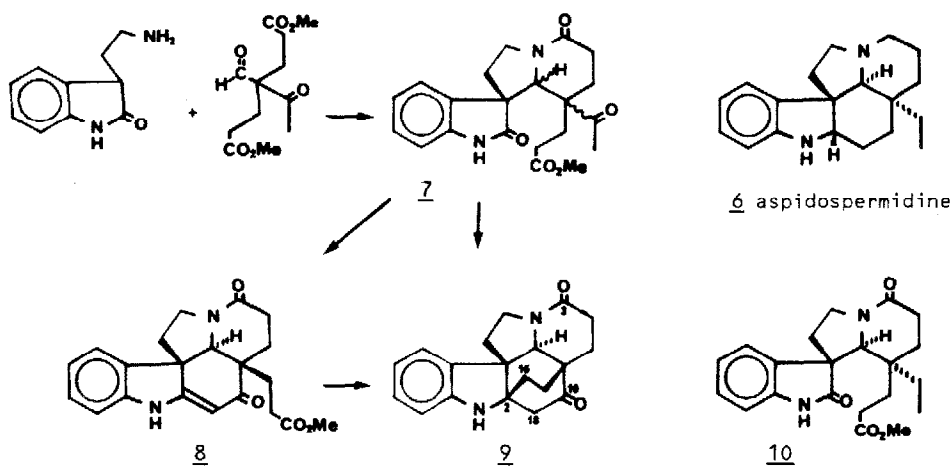
The intra-molecular cyclization of a methylketone onto the oxindolic lactam group (scheme 1) has been first developed by Ban<sup>1</sup> : he converted 1a to an iminoether (2a), which proved suitable for a base-catalyzed (NaH/DMSO) cyclization to 3. Later on, we were able to extend the process to some oxindolic methyl esters 1b, which were derived to iminoethers 2b (R=Me,Et) and further cyclized to anilinoacrylic esters 4 (R=Me,Et)<sup>2</sup>. The observed transesterification and a side-reaction of the dimsyl anion as well had led us to modify Ban's conditions by using a methyl iminoether and NaH/DMF as a basic medium. While the mechanism of this cyclization is not yet fully understood<sup>3</sup>, it allowed completion of short total syntheses in the indole alkaloid field<sup>2,4a-c</sup>.



Scheme 1

A striking improvement resulted when we heated esters 1b in PPA, which directly led to indolenines 5 upon cyclization and simultaneous loss of CO<sub>2</sub>Me. This last reagent allowed a synthesis of ( $\pm$ )-aspidospermidine 6

in only three steps from 2-hydroxytryptamine<sup>2</sup> and, still more dramatically, a synthesis of the hexacyclic 3,19-dioxoaspidofractinine **9** in two steps from 2-hydroxytryptamine<sup>5</sup> (scheme 2).

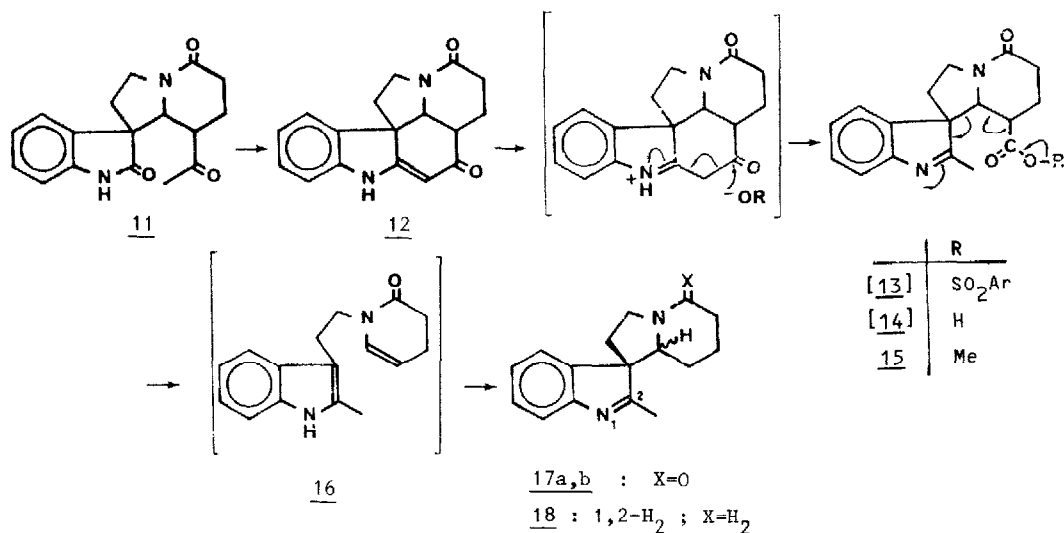


**Scheme 2**

It remained unclear which of the two carbons C-16 or C-18 had first undergone the cyclization onto C-2. We have now observed that para-toluene-sulfonic acid in boiling toluene (15 h) also performs the transformation of **7** to **9** (Yield : 50% from the more polar isomer or 23% from 2-hydroxytryptamine), while it does not affect the oxindolic ester **10**. Moreover, compound **8**<sup>10</sup> was prepared from **7** ( i/  $\text{Me}_3\text{OBF}_4$  ; ii/  $\text{NaH}$ , DMF, 48 h, 20°C, 52%) and further heated with TsOH in toluene to yield the hexacyclic ketolactam **9** (c.a. 20%). Thus, the double cyclization of **7** to **9** (TsOH) is thought to proceed through **8**, which involves a direct cyclization of an oxindolic methylketone (**1a**, scheme 1) to an enaminketone **3**.

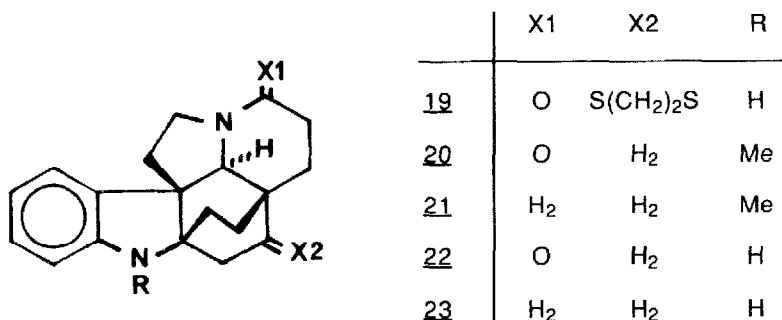
These results prompted to test the reactivity of the model oxindolic methylketone **11**<sup>6,7</sup> towards TsOH (scheme 3). Reaction of **11** (major, more polar isomer) with TsOH in boiling toluene yielded three derivatives, which were enaminketone **12** (traces) and the two epimeric indolenines **17a,b** (14% ; **17a**, less polar/**17b**, more polar – 5/7). Reduction of **17a** with  $\text{LiAlH}_4$  in THF gave the methylindoline **18** (methyl doublet at 1.30 ppm,  $J=7.2$  Hz in  $^1\text{H}$  NMR). The structure of **12** was supported by its synthesis from **11** using Ban's iminoether procedure (27%). Furthermore, upon heating in toluene plus TsOH, **12** again generated **17a** and **17b**. Interesting, when MeOH was added to the reaction mixture, the methyl ester **15** resulted (58%). Moreover, **11** was directly transformed into **15** (11%), along with **17a,b** (12%) upon treatment with TsOH in toluene, addition of MeOH and reflux for 1 h.

These facts fit with the succession of events depicted on scheme 3, i.e. cyclization of **11** to **12**, fragmentation of the protonated **12** to a mixed anhydride **13**, which yields either acid **14** ( $\text{H}_2\text{O}$ ) or ester **15** (MeOH). Acid **14** itself decarboxylates in acidic medium (arrows) to enamide **16**, which finally cyclizes to **17a,b**. The poor 14% yield in the reaction probably reflects incomplete cyclization to **12** and decarboxylation of **14** under the reaction conditions used. The initial "activation" of the lactam carbonyl group by tosylic acid (and by PPA as well) still awaits further experiments to be fully understood<sup>8</sup>.



Scheme 3

Finally, the hexacyclic ketolactam **9** was sequentially reduced in order to complete the total synthesis of ( $\pm$ )-aspidofractinine **23**: hydrogenolysis of the thioether **19** (ethanedithiol, AcOH, BF<sub>3</sub>, r.t. 12 h, 70%) with Raney nickel in MeOH exclusively gave the N(1)-methylactam **20** (98%), which was reduced with LiAlH<sub>4</sub> to ( $\pm$ )-N-methyl-aspidofractinine **21** (84%). Replacing MeOH by EtOH in the desulfuration step gave 3-oxoaspidofractinine **22** (87%), which was reduced by LiAlH<sub>4</sub> to ( $\pm$ )-aspidofractinine **23** (70%)<sup>9,10</sup>.

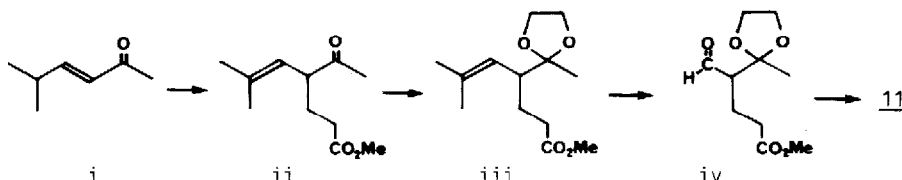


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#### References and notes :

1. T. Oishi, M. Nagai and Y. Ban, *Tetrahedron Lett.*, 1968, p.491.
2. J.-Y. Laronze, J. Laronze-Fontaine, J. Lévy and J. Le Men, *Tetrahedron Lett.*, 1974, p.491.
3. It probably involves participation of the other fonctionnal groups present in the molecule.
4. a) J. Lévy, J.-Y. Laronze, J. Laronze and J. Le Men, *Tetrahedron Lett.*, 1978, p.1579.  
 b) J.-Y. Laronze, D. Cartier, J. Laronze and J. Lévy, *Tetrahedron Lett.*, 1980, p.4441.  
 c) D. Cartier, M. Ouahrani, G. Hugel and J. Lévy, *Heterocycles*, 1988, **27**, 657.
5. a) D. Cartier, D. Patigny and J. Lévy, *Tetrahedron Lett.*, 1982, p.1897.  
 b) D. Cartier and J. Lévy, in press.

6. Y. Ban, T. Ohnuma, M. Nagai, Y. Sendo and T. Oishi, *Tetrahedron Lett.*, 1982, p.5023.  
 7. While this compound had been previously synthesised by Ban<sup>6</sup>, a more convergent access was devised as follows: 5-methyl-4-hexene-2-one (i) was monoalkylated with methyl acrylate to (ii)<sup>5a,b</sup> and further conducted to dioxolane(iii), which was ozonized to (iv). Condensation of (iv) with 2-hydroxytryptamine (1/ PhH, reflux with a Dean-Stark apparatus ; 2/ AcOH, reflux, 20h) yielded **11** (70%) as a mixture of (racemic) isomers, which were separated (tlc) into three compounds in the ratio 15:35:50, in order of increasing polarities.



8. It may be pointed out that the lactam carbonyl of oxindoles, like that in anilides and in N-acylindoles, partly behave as a ketone towards reducing agents.  
 9. The synthetic aspidofractinine was compared with an authentic sample kindly provided by Professor Le Men-Olivier. *Isolation and structure of aspidofractinine* : C. Djerassi, H. Budzikiewicz, R.J. Owellen, J.M. Wilson, W.G. Kump, D.J. Le Count, A.R. Battersby and H. Schmid, *Helv. Chim. Acta*, 1963, **46**, 742. B.W. Bycroft, D. Schumann, M.B. Patel and H. Schmid, *Helv. Chim. Acta*, 1964, **47**, 1147. *Previous syntheses of aspidofractinine* : A. Guggisberg, A.A. Gorman, B.W. Bycroft and H. Schmid, *Helv. Chim. Acta*, 1969, **52**, 76. Y. Ban, Y. Honma and T. Oishi, *Tetrahedron Lett.*, 1976, p.1111.  
 10. The structures of the new compounds were fully consistent with their UV, IR, MS, <sup>1</sup>H and <sup>13</sup>C NMR spectra. For examples :

	M S M <sup>+</sup> : (%), base peak	<sup>1</sup> H N M R (300 MHz) δ (ppm), J (Hz)
<b>15</b>	312(12), 157	7.58(1H,d, J=7.6) 7.38(1H,dd,J <sub>1</sub> =J <sub>2</sub> =7.6) 7.20(1H,dd,J <sub>1</sub> =J <sub>2</sub> =7.6) 7.03(1H,d, J=7.6) 4.25(1H,d,J=9) 4.04(1H,m) 3.86(1H,m) 3.45(3H,s) 2.53-2.38(3H,m) 2.30(3H,s) 1.95-1.85(4H,m)
<b>17a</b>	254(13), 157	7.55(1H,d, J=7.6) 7.37(1H,m) 7.25(2H,m) 4.08(1H,m) 3.96(1H,dd,J <sub>1</sub> =4.5,J <sub>2</sub> =11.2) 2.5-2.3(3H,m) 2.25(3H,s) 2.10(1H,m) 1.90(1H,m) 1.63(1H,m) 1.40(1H,m) 1.04(1H,dt,J <sub>1</sub> =4.5,J <sub>2</sub> =11.2)
<b>17b</b>	254(13), 157	7.58(1H,d, J=7) 7.36(1H,dd,J <sub>1</sub> =J <sub>2</sub> =7.6) 7.18(1H,dd,J <sub>1</sub> =J <sub>2</sub> =7) 7.04(1H,d, J=7.6) 4.09(1H,m) 3.97(1H,dd,J <sub>1</sub> =4.5,J <sub>2</sub> =11.2) 3.84(1H,m) 2.41(2H,m) 2.32(3H,s) 2.26(1H,m) 1.80(1H,m) 1.68(1H,m) 1.37(1H,m) 0.75(1H,dt,J <sub>1</sub> =4.5,J <sub>2</sub> =11.2)
<b>18</b>	242(7), 97	7.05(1H,dd,J <sub>1</sub> =J <sub>2</sub> =8) 7.03(1H,dd,J <sub>1</sub> =J <sub>2</sub> =7.6) 6.78(1H,d, J=7.6) 6.60(1H,dd,J <sub>1</sub> =J <sub>2</sub> =8) 3.82(1H,q,J=7.2) ...1.30(3H,d,J=7.2)
<b>19</b>	384(98), 169	7.08(2H,dd,J <sub>1</sub> =J <sub>2</sub> =8) 6.77(1H,dd,J <sub>1</sub> =J <sub>2</sub> =8) 6.66(1H,d, J=8) 4.18(1H,dd,J <sub>1</sub> =8.3,J <sub>2</sub> =12) 3.92(1H,d,J=1.5) 3.38-3.05(5H,m) 2.66(1H,dd,J <sub>1</sub> =3,J <sub>2</sub> =14.5) 2.58-1.62(10H,m) 1.48(1H,dd,J <sub>1</sub> =6,J <sub>2</sub> =14) 1.40(1H,m)
<b>20</b>	308(37), 170	7.14(1H,dd,J <sub>1</sub> =J <sub>2</sub> =8) 7.07(1H,d,J=8) 6.72(1H,dd,J <sub>1</sub> =J <sub>2</sub> =8) 6.46(1H,d, J=8) 4.31(1H,dd,J <sub>1</sub> =7.5,J <sub>2</sub> =12) 3.59(1H,d,J=1) 3.15(1H,ddd,J <sub>1</sub> =5.2,J <sub>2</sub> =J <sub>3</sub> =12) 2.60(3H,s) 2.38(2H,m) 2.10(1H,ddd,J <sub>1</sub> =7.5,J <sub>2</sub> =J <sub>3</sub> =12)-1.95-1.30(12H,m)
<b>21</b>	294(84), 109	7.36(1H,d,J=8) 7.06(1H,dd,J <sub>1</sub> =J <sub>2</sub> =8) 6.74(1H,dd,J <sub>1</sub> =J <sub>2</sub> =8) 6.40(1H,d, J=8) 3.23(1H,dd,J <sub>1</sub> =8,J <sub>2</sub> =17) 3.15-3.00(3H,m) 2.70(1H,ddd,J <sub>1</sub> =3.7,J <sub>2</sub> =8,J <sub>3</sub> =14) 2.56(3H,s) 2.20-1.20(13H,m)
<b>22</b>	294(72), 169	7.15(1H,d,J=7.6) 7.08(1H,dd,J <sub>1</sub> =J <sub>2</sub> =7.6) 6.83(1H,dd,J <sub>1</sub> =J <sub>2</sub> =7.6) 6.75(1H,d, J=7.6) 4.34(1H,dd,J <sub>1</sub> =8,J <sub>2</sub> =11) 3.80(1H,s) 3.30(1H,ddd,J <sub>1</sub> =5.4,J <sub>2</sub> =8,J <sub>3</sub> =11) 2.70-1.35 (15H,m)